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1-Phenyl-2-trifluoromethyl-4-quinolone

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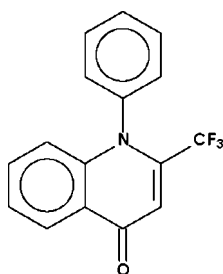
Received 22 November 2007; accepted 24 November 2007

Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.041; wR factor = 0.147; data-to-parameter ratio = 15.5.

In the title molecule, $\text{C}_{16}\text{H}_{10}\text{F}_3\text{NO}$, the N-bound phenyl ring is oriented nearly orthogonal to the quinolinyl ring in order to avoid steric clashes with the trifluoromethyl substituent [dihedral angle 89.7 (1°)].

Related literature

For synthesis, see: Sosnovskikh *et al.* (2005); Usachev & Sosnovskikh (2004).



Experimental

Crystal data

 $\text{C}_{16}\text{H}_{10}\text{F}_3\text{NO}$ $M_r = 289.25$ Monoclinic, $P2_1/c$
 $a = 8.7403$ (5) Å
 $b = 17.574$ (1) Å
 $c = 8.7559$ (6) Å
 $\beta = 103.931$ (2) $^\circ$
 $V = 1305.4$ (1) Å³
 $Z = 4$

Mo $K\alpha$ radiation
 $\mu = 0.12$ mm⁻¹
 $T = 295$ (2) K
 $0.35 \times 0.25 \times 0.15$ mm

Data collection

Rigaku R-Axis RAPID
diffractometer
Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)
 $T_{\min} = 0.843$, $T_{\max} = 0.982$

20310 measured reflections
2970 independent reflections
1882 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.030$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.147$
 $S = 1.14$
2970 reflections

191 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.22$ e Å⁻³
 $\Delta\rho_{\min} = -0.23$ e Å⁻³

Data collection: *RAPID-AUTO* (Rigaku Corporation, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2008).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2227).

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supporting information

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1-Phenyl-2-trifluoromethyl-4-quinolone

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S1. Comment

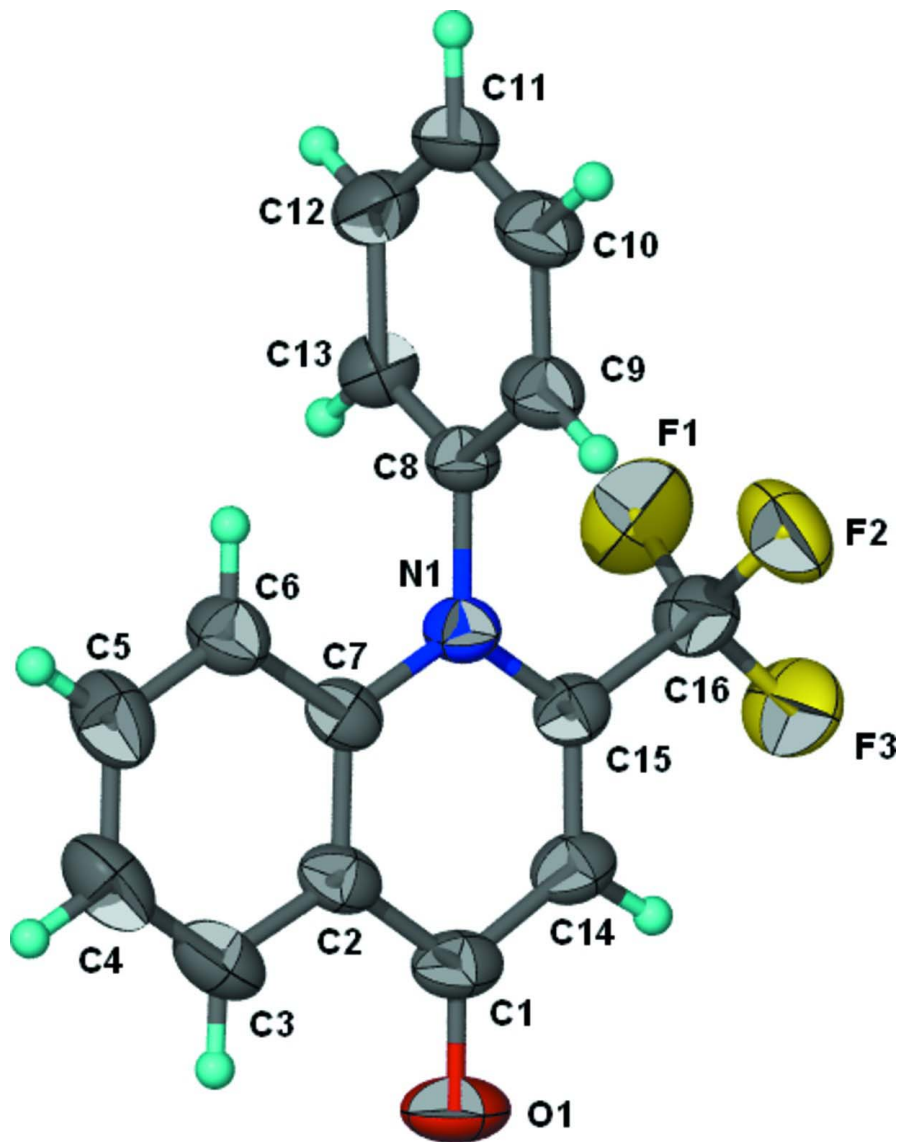
Compound (I) was isolated during an attempted reaction of the corresponding thione with CuCl₂ in THF solution, see Experimental. The N-bound aromatic ring in (I) occupies a position orthogonal to the quinolinyl ring so as to avoid steric clashes with the adjacent CF₃ group, Fig. 1.

S2. Experimental

The synthesis of (I) has been described by Usachev & Sosnovskikh (2004); also see Sosnovskikh *et al.* (2005). In the present study, (I) was obtained as a side-product when the thione was recrystallized from THF in the presence of copper(II) chloride.

S3. Refinement

Carbon-bound H atoms were included in the refinement in the riding-model approximation with C—H = 0.93 Å, and with $U_{\text{iso}}(\text{H})$ 1.2 $U_{\text{eq}}(\text{C})$.

**Figure 1**

Molecular structure of (I) showing displacement ellipsoids at the 50% probability level and H atoms as spheres of arbitrary radius.

1-Phenyl-2-trifluoromethyl-4-quinolone

Crystal data

$C_{16}H_{10}F_3NO$

$M_r = 289.25$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 8.7403\ (5)\ \text{\AA}$

$b = 17.574\ (1)\ \text{\AA}$

$c = 8.7559\ (6)\ \text{\AA}$

$\beta = 103.931\ (2)^\circ$

$V = 1305.4\ (1)\ \text{\AA}^3$

$Z = 4$

$F(000) = 592$

$D_x = 1.472\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 12413 reflections

$\theta = 3.2\text{--}27.5^\circ$

$\mu = 0.12\ \text{mm}^{-1}$

$T = 295\ \text{K}$

Prism, yellow

$0.35 \times 0.25 \times 0.15\ \text{mm}$

Data collection

Rigaku R-Axis RAPID
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 10.000 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.843$, $T_{\max} = 0.982$

20310 measured reflections

2970 independent reflections

1882 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.030$

$\theta_{\max} = 27.4^\circ$, $\theta_{\min} = 3.2^\circ$

$h = -11 \rightarrow 10$

$k = -22 \rightarrow 22$

$l = -11 \rightarrow 11$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.147$

$S = 1.14$

2970 reflections

191 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0722P)^2 + 0.1288P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$

Extinction correction: *SHELXL97* (Sheldrick,
1997), $F_c^* = kF_c[1 + 0.001x\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.005 (3)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.43638 (17)	0.24350 (8)	0.2257 (2)	0.1064 (6)
F2	0.43881 (16)	0.34786 (9)	0.10584 (14)	0.0941 (5)
F3	0.65236 (14)	0.30060 (9)	0.23970 (17)	0.0969 (5)
O1	0.72788 (16)	0.43177 (9)	0.73522 (17)	0.0794 (5)
N1	0.33546 (14)	0.36764 (7)	0.39971 (15)	0.0456 (3)
C1	0.6081 (2)	0.41335 (10)	0.6337 (2)	0.0560 (5)
C2	0.4482 (2)	0.42888 (9)	0.65010 (19)	0.0496 (4)
C3	0.4245 (2)	0.46671 (12)	0.7840 (2)	0.0660 (5)
H3	0.5114	0.4821	0.8618	0.079*
C4	0.2772 (3)	0.48118 (13)	0.8018 (2)	0.0768 (6)
H4	0.2639	0.5067	0.8907	0.092*
C5	0.1465 (3)	0.45791 (13)	0.6873 (2)	0.0721 (6)
H5	0.0458	0.4676	0.7004	0.086*
C6	0.1646 (2)	0.42053 (11)	0.5544 (2)	0.0590 (5)
H6	0.0764	0.4051	0.4781	0.071*
C7	0.31618 (19)	0.40578 (9)	0.53450 (18)	0.0461 (4)
C8	0.19504 (17)	0.34533 (9)	0.28127 (19)	0.0460 (4)
C9	0.1299 (2)	0.39618 (11)	0.1639 (2)	0.0557 (4)
H9	0.1768	0.4433	0.1582	0.067*
C10	-0.0070 (2)	0.37585 (13)	0.0544 (2)	0.0655 (5)
H10	-0.0520	0.4092	-0.0264	0.079*
C11	-0.0764 (2)	0.30637 (14)	0.0650 (2)	0.0701 (6)
H11	-0.1682	0.2931	-0.0086	0.084*

C12	−0.0110 (2)	0.25642 (13)	0.1837 (3)	0.0691 (6)
H12	−0.0589	0.2097	0.1904	0.083*
C13	0.1267 (2)	0.27569 (11)	0.2936 (2)	0.0575 (5)
H13	0.1719	0.2422	0.3741	0.069*
C14	0.61571 (19)	0.37504 (10)	0.4918 (2)	0.0547 (4)
H14	0.7140	0.3646	0.4733	0.066*
C15	0.48470 (18)	0.35353 (9)	0.3836 (2)	0.0472 (4)
C16	0.5026 (2)	0.31135 (12)	0.2388 (2)	0.0626 (5)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.1113 (11)	0.0740 (9)	0.1504 (14)	−0.0158 (7)	0.0641 (10)	−0.0509 (9)
F2	0.0938 (10)	0.1323 (13)	0.0570 (8)	0.0227 (8)	0.0197 (7)	−0.0097 (7)
F3	0.0547 (7)	0.1408 (13)	0.1002 (10)	0.0159 (7)	0.0283 (6)	−0.0278 (9)
O1	0.0612 (8)	0.0894 (10)	0.0712 (10)	−0.0143 (7)	−0.0163 (7)	−0.0032 (8)
N1	0.0397 (7)	0.0496 (7)	0.0446 (8)	0.0023 (5)	0.0043 (5)	−0.0038 (6)
C1	0.0524 (10)	0.0504 (9)	0.0553 (10)	−0.0051 (7)	−0.0062 (8)	0.0093 (8)
C2	0.0587 (10)	0.0425 (8)	0.0412 (8)	−0.0009 (7)	−0.0004 (7)	0.0039 (7)
C3	0.0828 (14)	0.0647 (12)	0.0432 (10)	−0.0016 (9)	0.0010 (9)	−0.0027 (9)
C4	0.1019 (16)	0.0793 (14)	0.0482 (11)	0.0114 (12)	0.0165 (11)	−0.0099 (10)
C5	0.0777 (13)	0.0850 (14)	0.0563 (12)	0.0204 (11)	0.0217 (10)	−0.0009 (10)
C6	0.0551 (10)	0.0703 (12)	0.0495 (10)	0.0102 (8)	0.0088 (8)	−0.0035 (9)
C7	0.0514 (9)	0.0436 (8)	0.0405 (8)	0.0058 (6)	0.0054 (7)	0.0028 (7)
C8	0.0382 (8)	0.0549 (9)	0.0427 (9)	0.0013 (6)	0.0056 (6)	−0.0053 (7)
C9	0.0509 (9)	0.0630 (11)	0.0500 (10)	0.0054 (7)	0.0057 (7)	−0.0019 (8)
C10	0.0516 (10)	0.0912 (15)	0.0481 (10)	0.0131 (10)	0.0013 (8)	−0.0037 (10)
C11	0.0439 (9)	0.1105 (18)	0.0526 (11)	−0.0076 (10)	0.0052 (8)	−0.0243 (11)
C12	0.0532 (10)	0.0841 (14)	0.0708 (13)	−0.0185 (9)	0.0163 (9)	−0.0180 (11)
C13	0.0509 (9)	0.0637 (11)	0.0569 (10)	−0.0061 (8)	0.0111 (8)	−0.0016 (8)
C14	0.0406 (9)	0.0570 (10)	0.0621 (11)	0.0011 (7)	0.0039 (7)	0.0049 (8)
C15	0.0428 (8)	0.0460 (9)	0.0510 (10)	0.0028 (6)	0.0081 (7)	0.0027 (7)
C16	0.0502 (10)	0.0700 (12)	0.0682 (12)	0.0045 (9)	0.0157 (8)	−0.0098 (10)

Geometric parameters (Å, °)

F1—C16	1.318 (2)	C6—C7	1.402 (2)
F2—C16	1.328 (2)	C6—H6	0.9300
F3—C16	1.321 (2)	C8—C9	1.377 (2)
O1—C1	1.2421 (19)	C8—C13	1.377 (2)
N1—C15	1.368 (2)	C9—C10	1.389 (2)
N1—C7	1.402 (2)	C9—H9	0.9300
N1—C8	1.4569 (19)	C10—C11	1.376 (3)
C1—C14	1.429 (3)	C10—H10	0.9300
C1—C2	1.464 (3)	C11—C12	1.375 (3)
C2—C7	1.400 (2)	C11—H11	0.9300
C2—C3	1.405 (3)	C12—C13	1.390 (3)
C3—C4	1.358 (3)	C12—H12	0.9300

C3—H3	0.9300	C13—H13	0.9300
C4—C5	1.388 (3)	C14—C15	1.353 (2)
C4—H4	0.9300	C14—H14	0.9300
C5—C6	1.378 (3)	C15—C16	1.509 (3)
C5—H5	0.9300		
C15—N1—C7	118.97 (13)	C8—C9—C10	118.76 (19)
C15—N1—C8	122.58 (13)	C8—C9—H9	120.6
C7—N1—C8	118.45 (12)	C10—C9—H9	120.6
O1—C1—C14	122.49 (18)	C11—C10—C9	120.14 (19)
O1—C1—C2	122.78 (18)	C11—C10—H10	119.9
C14—C1—C2	114.73 (14)	C9—C10—H10	119.9
C7—C2—C3	118.62 (17)	C10—C11—C12	120.51 (17)
C7—C2—C1	121.02 (16)	C10—C11—H11	119.7
C3—C2—C1	120.36 (16)	C12—C11—H11	119.7
C4—C3—C2	121.24 (18)	C11—C12—C13	120.06 (19)
C4—C3—H3	119.4	C11—C12—H12	120.0
C2—C3—H3	119.4	C13—C12—H12	120.0
C3—C4—C5	120.03 (19)	C8—C13—C12	118.78 (18)
C3—C4—H4	120.0	C8—C13—H13	120.6
C5—C4—H4	120.0	C12—C13—H13	120.6
C6—C5—C4	120.56 (19)	C15—C14—C1	122.15 (16)
C6—C5—H5	119.7	C15—C14—H14	118.9
C4—C5—H5	119.7	C1—C14—H14	118.9
C5—C6—C7	119.82 (18)	C14—C15—N1	122.97 (16)
C5—C6—H6	120.1	C14—C15—C16	118.99 (15)
C7—C6—H6	120.1	N1—C15—C16	118.04 (14)
C2—C7—N1	120.14 (15)	F1—C16—F3	106.56 (16)
C2—C7—C6	119.74 (15)	F1—C16—F2	106.15 (17)
N1—C7—C6	120.13 (14)	F3—C16—F2	106.31 (17)
C9—C8—C13	121.74 (16)	F1—C16—C15	112.65 (16)
C9—C8—N1	118.85 (15)	F3—C16—C15	111.59 (15)
C13—C8—N1	119.34 (15)	F2—C16—C15	113.10 (16)
O1—C1—C2—C7	−178.65 (16)	C13—C8—C9—C10	−0.9 (3)
C14—C1—C2—C7	1.2 (2)	N1—C8—C9—C10	−177.78 (15)
O1—C1—C2—C3	0.6 (3)	C8—C9—C10—C11	0.7 (3)
C14—C1—C2—C3	−179.60 (16)	C9—C10—C11—C12	−0.1 (3)
C7—C2—C3—C4	−0.3 (3)	C10—C11—C12—C13	−0.4 (3)
C1—C2—C3—C4	−179.56 (19)	C9—C8—C13—C12	0.5 (3)
C2—C3—C4—C5	0.6 (3)	N1—C8—C13—C12	177.32 (16)
C3—C4—C5—C6	−0.5 (3)	C11—C12—C13—C8	0.2 (3)
C4—C5—C6—C7	0.1 (3)	O1—C1—C14—C15	178.37 (17)
C3—C2—C7—N1	−179.75 (15)	C2—C1—C14—C15	−1.4 (2)
C1—C2—C7—N1	−0.5 (2)	C1—C14—C15—N1	1.1 (3)
C3—C2—C7—C6	−0.1 (3)	C1—C14—C15—C16	−178.30 (16)
C1—C2—C7—C6	179.14 (15)	C7—N1—C15—C14	−0.3 (2)
C15—N1—C7—C2	0.0 (2)	C8—N1—C15—C14	178.98 (15)

C8—N1—C7—C2	−179.28 (14)	C7—N1—C15—C16	179.08 (15)
C15—N1—C7—C6	−179.62 (15)	C8—N1—C15—C16	−1.6 (2)
C8—N1—C7—C6	1.1 (2)	C14—C15—C16—F1	120.25 (19)
C5—C6—C7—C2	0.2 (3)	N1—C15—C16—F1	−59.1 (2)
C5—C6—C7—N1	179.87 (17)	C14—C15—C16—F3	0.4 (3)
C15—N1—C8—C9	−91.06 (19)	N1—C15—C16—F3	−178.97 (16)
C7—N1—C8—C9	88.22 (19)	C14—C15—C16—F2	−119.37 (18)
C15—N1—C8—C13	92.0 (2)	N1—C15—C16—F2	61.2 (2)
C7—N1—C8—C13	−88.73 (19)		
